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Determination of the Absolute Configuration of Yessotoxin, a Polyether Compound Implicated in Diarrhetic Shellfish Poisoning, by NMR Spectroscopic Method Using a Chiral Anisotropic Reagent, Methoxy-(2-naphthyl)acetic Acid

Haruko Takahashi and Takenori Kusumi*

Faculty of Pharmaceutical Sciences, The University of Tokushima, Tokushima 770, Japan

Yukiko Kan

Faculty of Pharmaceutical Sciences, Tokushima Bunri University, Tokushima 770, Japan

Masayuki Satake and Takeshi Yasumoto*

Faculty of Agriculture, Tohoku University, Tsutsumidori-Amamiya, Sendai 981, Japan

Abstract: The absolute configuration of yessotoxin (1), a potent diarrhetic toxin causative of shellfish poisoning, was determined by NMR-spectroscopy using a chiral anisotropic reagent, methoxy-(2-naphthyl)acetic acid (2NMA), which was applied to a monoacetate of bisdesulfated yessotoxin (3). Copyright © 1996 Elsevier Science Ltd

A significant number of polyethers, most of them having potential biological activities, has been isolated from marine organisms.¹ Their intriguing structures have been investigated mainly by NMR spectroscopy and X-ray crystallography. As they are present in extremely minute amounts in the organisms, however, the stereochemistry of their structures are proposed relatively, except for very few examples such as brevetoxins, the absolute configuration of which has been established by CD.²

Yessotoxin (1: YTX) is a brevetoxin-type polyether that has been isolated from scallops as a diarrhetic shellfish poison.³ Its planar structure and the relative stereochemistry have been elucidated by spectroscopic methods, although the absolute configuration has remained undetermined.

We have been developing new chiral anisotropic reagents, such as 2NMA [methoxy-(2-naphthyl)acetic acid], 4.5 9ATMA (9-anthrylmethoxyacetic acid), 4.5 2ATMA (2-anthrylmethoxyacetic acid), 6 and PGME (phenylglycine methyl ester), 7 which make it possible to determine the absolute configurations of organic compounds by NMR. Among these reagents, 2NMA is the most convenient for

secondary alcohols in that (i) the anisotropy of its naphthalene ring is much greater than that of the benzene ring of MTPA (conventional modified Mosher's method⁸), giving greater $\Delta\delta$ (δ_R - δ_S) values, and (ii) the enantiomerically pure reagent is easily prepared in substantial amounts.⁹ This report deals with determination of the absolute configuration of YTX by means of 2NMA.

Attempted esterification at 32-OH of YTX with 2NMA was unsuccessful because of its axial orientation. Therefore, we focused on introducing 2NMA at 4-position, which is blocked by a sulfate group. Solvolysis of YTX was facilitated by heating a solution of YTX in pyridine-dioxane (1:1, 120 °C, 3 h) to give bisdesulfated YTX (2: DS-YTX). When DS-YTX (1.6 mg) was treated with an excess of acetic anhydride (50 μ L) in pyridine (50 μ L) at 0 °C for 45 min, a mixture of recovered DS-YTX (300 μ g), monoacetate (3: 600 μ g), ¹⁰ and diacetate (4: 300 μ g), separable by HPLC (Mightysil RP-18; MeOH: MeCN: H₂O = 9: 1: 1), was obtained. The secondary OH at C-32 was intact in 3 and 4.

Monoacetate 3 having free OH at C-4 was divided in two portions (300 μ g each), and each portion was treated with (R)- and (S)-2NMA (1.1 mg), EDC¹¹ (3.9 mg), DMAP (1.5 mg), and NEt₃ (2.8 μ L) in chloroform (500 μ L) for 43.5 h at ambient temperature. The reaction mixture was successively washed with 10% aq citric acid, H₂O, 5% aq NaHCO₃, and H₂O, and the solvent was removed. Each residue was purified by HPLC (see above) to give pure (R)- (5: 200 μ g) and (S)-2NMA esters (6: 200 μ g), respectively. The ¹H-NMR spectra of 3, 5, and 6 are shown in Fig. 1.

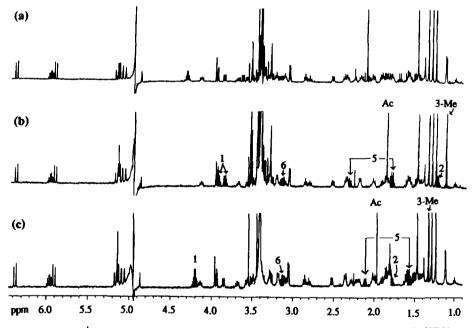


Figure 1. A part of the ¹H-NMR spectra (600 MHz, CD₃OD) of (a) monoacetate (3), (b) monoacetyl-(S)-2NMA ester (6) and (c) monoacetyl-(R)-2NMA ester (5) of DS-YTX. Spectra were recorded using a presaturation technique irradiating at the strong OH bands at 4.95 ppm except during acquisition.

By analyzing the COSY and HOHAHA spectra of 5 and 6, much higher $\Delta\delta$ values $[\delta_{R(5)} - \delta_{S(6)}]$ of the protons with much greater magnitude than those anticipated for MTPA were obtained. They are indicated in partial structure 7.

From these results, the absolute configuration of the hydroxy group at C-4 was determined to be S, and, since the relative stereochemistry of the other asymmetric centers has been established, the present finding led to the absolute configuration of YTX as shown in 1.

This is the second report (the first one on brevetoxin B²) on the absolute configuration of a ladderlike polyether, and the present methodology is expected to be applicable to the stereochemical as well as biosynthetic studies on other important polyethers.

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References and Notes

- 1. Yasumoto, T.; Murata, M. Chem. Rev., 1993, 93, 1897-1909.
- (a) Lin, Y.-Y.; Risk, M.; Ray, S. M.; Van Engen, D.; Clardy, J.; Golik, J.; James, J. C.; Nakanishi, K. J. Am. Chem. Soc., 1981, 103, 6773-6775. (b) Very recently, the absolute configuration of mitotoxin has been elucidated: (i) Sasaki, M.; Matsumori, N.; Maruyama, T.; Nonomura, T.; Murata, M.; Tachibana, K.; Yasumoto, T. Angew. Chem., Int. Ed. Engl., 1996, in press. (ii) Nonomura, T.; Sasaki, M.; Matsumori, N.; Murata, M.; Tachibana, K.; Yasumoto, T. Angew. Chem., Int. Ed. Engl., 1996, in press.
- 3. (a) Murata, M.; Kumagai, M.; Lee, J. S.; Yasumoto, T. Tetrahedron Lett., 1987, 28, 5869-5872. (b) Satake, M.; Terasawa, K.; Kadowaki, Y.; Yasumoto, T. Tetrahedron Lett., 1996, in press.
- 4. (a) Kusumi, T.; Takahashi, H.; Ping, X.; Fukushima, T.; Asakawa, Y.; Hashimoto, T.; Kan, Y.; Inouye, Y. *Tetrahedron Lett.*, **1994**, *35*, 4397-4400. (b) Kusumi, T.; Takahashi, H.; Hashimoto, T.; Kan, Y.; Asakawa, Y. *Chemistry Lett.*, **1994**, 1093-1094.
- (a) Seco, J. M.; Latypov, Sh. K.; Quinoa, E.; Riguera, R. Tetrahedron Lett., 1994, 35, 2921-2924.
 (b) Latypov, Sh. K.; Seco, J. M.; Quinoa, E.; Riguera, R. J. Org. Chem., 1995, 60, 504-515.
- 6. Kouda, K.; Kusumi, T.; Kan, Y.; Hashimoto, T.; Asakawa, Y. *Tetrahedron Lett.*, **1996**, 37, 4541-4544.
- 7. Nagai, Y.; Kusumi, T. Tetrahedron Lett., 1995, 36, 1853-1856.
- 8. Ohtani, I.; Kusumi, T.; Kashman, Y.; Kakisawa, H. J. Am. Chem. Soc., 1991, 113, 4092-4096.
- 9. (R)- and (S)-2NMA are available on request. (E-mail: tkusumi@ph.tokushima-u.ac. ip).
- 10. This sample is contaminated with less than 10% of another monoacetate (3': $R_1 = H$, $R_2 = Ac$). This minor product could be eliminated by HPLC after conversion into 2NMA esters.
- Kimura, T.; Takai, M.; Masui, Y.; Morikawa, T.; Sakakibara, S. *Biopolymers*, 1981, 20, 1823-1826.

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